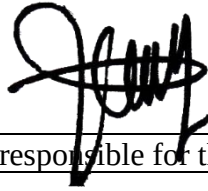
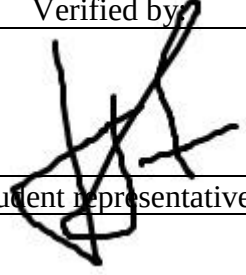




Faculty of Natural Sciences and Mathematics
Chemistry Department
Chemistry Education Study Program

Module name		Inorganic Chemistry Lab Work		
Module level, if applicable		2 nd year		
Code, if applicable		SPK – 322		
Semester(s) in which the module is taught		3 rd semester		
Person responsible for the module		M. Miqdam Musawwa, M.Sc.		
Lecturer		M. Miqdam Musawwa, M.Sc. Widinda Normalia Arlianty, M.Pd.		
Language		Indonesia		
Relation to curriculum		Compulsory		
Teaching methods	Class size	Forms of active participation	Workload 45 hours	
Practicum	20-25	Laboratory work, discussion	Laboratory work: 170 (min) x 8 (meeting)	23 hours
			Preparation: 120 (min) x 8 + 200 (min) Exam: 100 (min)+200 (min)	22 hours
ECTS credit		1.61		
Credit points		1 SCU		
Requirements according to the examination regulations		Student must follow all the series of practicum activities. Violation of this will result in giving an E value (failing practicum). Student who do not participate in the practicum for 3 (three) times without justified reasons may not attend the next practicum and are considered to have resigned from the practicum. Student who for some reason cannot follow the practicum according to the predetermined schedule can apply for inhal practicum. Inhal costs are determined by the laboratory. Student who inhal allowed for a student a maximum of 3 (three) times. Student who have not completed laboratory expenses such as tools, materials or tasks (if any) within a certain time will be given a K or F value.		
Recommended prerequisites		N/A		

Related course	Inorganic chemistry I dan Inorganic chemistry II		
Module objectives/intended learning outcomes	On successful completion of the course students should be able to: 1. Conduct practical activities in accordance with practical procedures. 2. Apply the basic principles of complex compound analysis 3. Apply stoichiometric concepts to complex compounds 4. Explain the concept of reaction of inorganic compound		
Content	• Determination of the molecular formula of a complex compound, • Preparation of complex salts and double salts, • Acid strength in aqueous medium, • Stabilization and isolation of Copper(I) compounds, • Making alum from waste aluminium foil and cans, • Stoichiometry of the reaction of Cu metal with Fe^{3+} salt		
Study and examination requirements and forms of examination	Final score (NA) is calculated as follows:		
	Intended learning outcomes	Weight (%)	Technique of assessment
	1	30	Non test: performance observation
	2	15	Non test: lab work report
	3	15	Non test: lab work report
	4	40	Test: pretest and posttest
Media employed	Inorganic chemistry laboratory equipment		
Reading list	Canham, G.R., 2000, Descriptive Inorganic Chemistry, Second edition, W.H. Housecroft, 2007, Inorganic Chemistry, (3rd Ed.), Prentice Hall. Scerri, E.R., 2006. The Periodic Table: Its Story and Its Significance, USA: Oxford University Press. Tim Penyusun, 2017, Buku Panduan Praktikum Kimia Anorganik, Yogyakarta: UII		

Prepared by:	Verified by:	Authorized by:
		
Person responsible for the module	Student representative	Coordinator Program